

Crystal structure of tris(morpholino)phosphonium triiodide, $(\text{OC}_4\text{H}_8\text{N})_3\text{PN}^+\text{H}_2\text{I}_3^-$

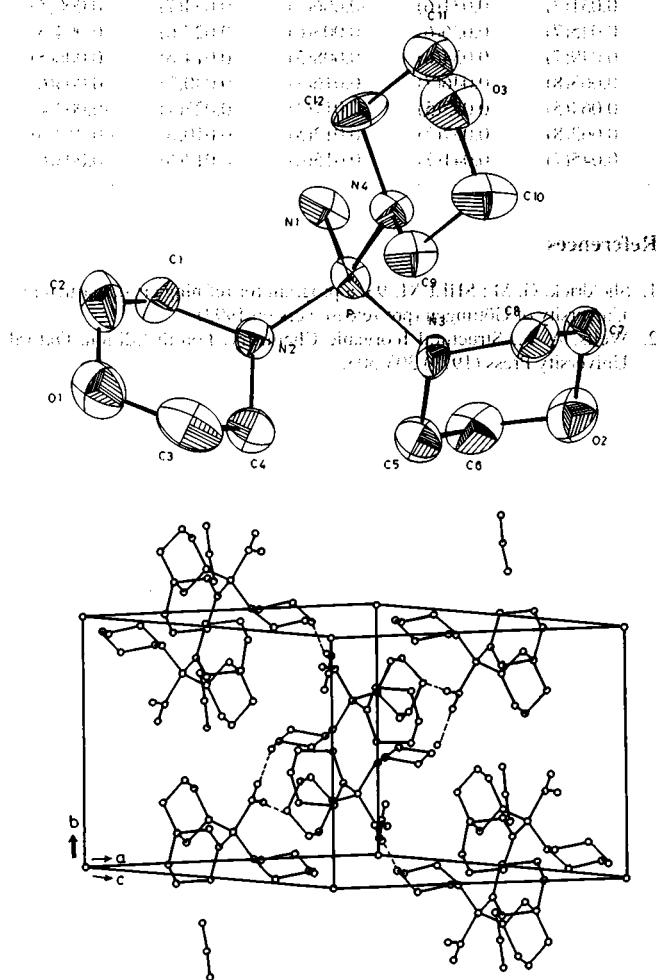
J. Srinivas, G. Sreenivasa Murthy

Indian Institute of Technology, Department of Physics, Madras 600 036, India

U. Swarnalatha and M. N. Sudheendra Rao

Indian Institute of Technology, Department of Chemistry, Madras 600 036, India

Received July 6, 1995, CSD-No. 402223



Source of material: The title compound was obtained from iodination reaction of $(OC_4H_8N)_3PN-S_3N_3$ with elemental iodine in acetonitrile medium and recrystallized from CH_3CN at room temperature.

The P–N distances ($d(\text{P}–\text{N}(1)) = 1.601 \text{ \AA}$, $d(\text{P}–\text{N}(2)) = 1.609 \text{ \AA}$, $d(\text{P}–\text{N}(3)) = 1.643 \text{ \AA}$, $d(\text{P}–\text{N}(4)) = 1.620 \text{ \AA}$) are considerably shorter than 1.77 \AA expected for normal $\text{P}(sp^3)–\text{N}(sp^2)$ bond, indicative of π -character in the bonds. The anion I_3^- is unsymmetrical and almost linear. The overall length of the anion (5.840 Å) is consistent with that observed in other triiodides (see ref. 2). N(1) takes part in two hydrogen bonds with O(2) and O(3) of two different fragments with N···O distances $2.89(1) \text{ \AA}$ and $2.98(1) \text{ \AA}$, respectively.

Qual é o desempenho da estratégia de apoio ao fundo? A estratégia é lucrativa?

Table 1. Parameters used for the X-ray data collection

Crystal: brown plate, size 0.08 x 0.22 x 0.30 mm
Wavelength: Mo K α radiation (0.71069 Å)
 μ : 44.28 cm $^{-1}$
Diffractometer: Enraf-Nonius CAD4
Scan mode: $\omega/2\theta$
T_{measurement}: 300 K
2θ_{max}: 50°
N(hkl)_{unique}: 2164
Criterion for I_0 : $I_0 > 3 \sigma(I_0)$
N(param)_{refined}: 312
Program: SHELXL-93

Table 2. Final atomic coordinates and displacement parameters (in Å²)

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso}
H(1A)	4e	0.097(7)	0.754(9)	0.095(6)	0.02(3)
H(1B)	4e	0.07(1)	0.65(1)	0.03(1)	0.10(5)
H(2A)	4e	-0.013(7)	0.685(9)	0.153(7)	0.03(3)
H(2B)	4e	-0.093(8)	0.713(9)	0.050(7)	0.05(3)
H(3A)	4e	0.022(6)	0.489(7)	0.200(6)	0.01(2)
H(3B)	4e	-0.02(1)	0.38(2)	0.152(9)	0.10(5)
H(4A)	4e	0.173(7)	0.427(8)	0.179(6)	0.03(3)
H(4B)	4e	0.139(9)	0.43(1)	0.085(8)	0.08(4)
H(5A)	4e	0.255(6)	0.867(7)	0.313(5)	0.00(2)
H(5B)	4e	0.15(1)	0.82(1)	0.324(8)	0.07(4)
H(6A)	4e	0.352(8)	0.836(9)	0.464(6)	0.03(3)
H(6B)	4e	0.250(9)	0.93(1)	0.454(8)	0.07(4)
H(7A)	4e	0.256(6)	0.562(8)	0.498(5)	0.02(2)
H(7B)	4e	0.38(1)	0.62(1)	0.497(9)	0.10(5)
H(8A)	4e	0.256(7)	0.521(9)	0.354(6)	0.03(3)
H(8B)	4e	0.161(9)	0.592(9)	0.345(7)	0.04(3)
H(9A)	4e	0.372(7)	0.374(9)	0.178(6)	0.04(3)
H(9B)	4e	0.305(6)	0.487(8)	0.101(6)	0.03(2)
H(10A)	4e	0.471(9)	0.57(1)	0.134(8)	0.06(4)
H(10B)	4e	0.48(1)	0.43(1)	0.12(1)	0.10(5)
H(11A)	4e	0.532(7)	0.647(9)	0.299(6)	0.03(2)
H(11B)	4e	0.612(9)	0.53(1)	0.355(7)	0.06(3)
H(12A)	4e	0.447(8)	0.43(1)	0.339(7)	0.06(3)
H(12B)	4e	0.433(9)	0.58(1)	0.353(8)	0.07(4)
HN(A')	4e	0.339(8)	0.84(1)	0.202(7)	0.05(3)
HN(B')	4e	0.285(8)	0.78(1)	0.115(8)	0.05(3)

Table 3. Final atomic coordinates and displacement parameters (in Å²)

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
I(1)	4e	0.0957(1)	0.0845(1)	0.11567(8)	0.1226(9)	0.0784(7)	0.0879(8)	0.0253(6)	0.0340(7)	0.0106(6)
I(2)	4e	0.23285(7)	0.17512(7)	0.28671(6)	0.0721(5)	0.0512(4)	0.0836(6)	0.0190(4)	0.0511(5)	0.0285(4)
I(3)	4e	0.36865(9)	0.2622(1)	0.46551(7)	0.1035(8)	0.0747(7)	0.0832(7)	0.0048(6)	0.0238(6)	0.0269(6)
P	4e	0.2687(2)	0.6526(2)	0.2184(2)	0.032(1)	0.034(1)	0.024(1)	-0.001(1)	0.015(1)	-0.000(1)
N(2)	4e	0.2694(6)	0.6906(7)	0.3172(5)	0.043(5)	0.036(4)	0.018(4)	0.005(3)	0.017(3)	-0.002(3)
N(3)	4e	0.3518(6)	0.5369(7)	0.2375(5)	0.027(4)	0.044(5)	0.028(4)	0.015(4)	0.008(4)	-0.008(4)
N(4)	4e	0.2957(7)	0.7773(8)	0.1762(6)	0.052(6)	0.040(5)	0.028(5)	-0.014(4)	0.020(4)	0.000(4)
N(1)	4e	0.1589(6)	0.6019(7)	0.1458(5)	0.036(5)	0.032(4)	0.030(4)	-0.002(3)	0.011(4)	0.001(3)
C(1)	4e	0.0745(9)	0.685(1)	0.0857(8)	0.045(7)	0.039(7)	0.045(7)	0.008(6)	0.010(5)	0.004(6)
C(2)	4e	-0.029(1)	0.667(1)	0.100(1)	0.034(7)	0.09(1)	0.056(9)	0.001(7)	0.012(6)	-0.006(8)
O(1)	4e	-0.0631(6)	0.5441(8)	0.0914(6)	0.041(5)	0.076(6)	0.072(6)	-0.014(4)	0.025(4)	-0.014(5)
C(3)	4e	0.014(1)	0.465(1)	0.152(1)	0.08(1)	0.072(9)	0.051(8)	-0.028(8)	0.043(8)	-0.021(8)
C(4)	4e	0.1204(9)	0.473(1)	0.1400(8)	0.039(6)	0.038(6)	0.051(7)	-0.008(5)	0.017(6)	0.003(5)
C(5)	4e	0.238(1)	0.814(1)	0.3372(7)	0.08(1)	0.034(6)	0.038(7)	0.017(6)	0.033(6)	0.002(5)
C(6)	4e	0.284(1)	0.840(1)	0.4338(8)	0.070(9)	0.056(7)	0.037(7)	0.005(7)	0.022(7)	-0.011(6)
O(2)	4e	0.2524(7)	0.7452(8)	0.4824(5)	0.090(6)	0.073(6)	0.036(4)	0.009(5)	0.039(4)	-0.003(4)
C(7)	4e	0.284(1)	0.626(1)	0.4645(7)	0.09(1)	0.051(7)	0.031(6)	-0.006(7)	0.034(7)	0.000(5)
C(8)	4e	0.235(1)	0.597(1)	0.3673(7)	0.058(8)	0.045(7)	0.036(6)	-0.003(6)	0.027(6)	0.004(5)
C(9)	4e	0.3648(9)	0.464(1)	0.1641(7)	0.054(7)	0.049(7)	0.034(6)	0.008(5)	0.014(6)	-0.006(5)
C(10)	4e	0.469(1)	0.493(1)	0.1542(8)	0.064(8)	0.063(8)	0.046(7)	0.016(7)	0.040(7)	-0.004(6)
O(3)	4e	0.5572(6)	0.4729(7)	0.2338(5)	0.048(5)	0.064(5)	0.053(5)	0.017(4)	0.027(4)	-0.003(4)
C(11)	4e	0.5481(9)	0.551(1)	0.3004(8)	0.031(6)	0.062(8)	0.057(7)	0.012(5)	0.010(6)	-0.012(6)
C(12)	4e	0.4475(9)	0.525(1)	0.3189(8)	0.059(7)	0.045(7)	0.041(7)	0.015(6)	0.017(6)	0.003(6)

Acknowledgment. We are thankful to the Head, Regional Sophisticated Instrumentation Centre, Indian Institute of Technology, Madras for the use of Enraf-Nonius CAD4 instrument.

References

- Sheldrick, G. M.: SHELXL-93, a program for refining crystal structures. University of Göttingen (pre-release version 1992).
- Wells, A. F.: Structural Inorganic Chemistry. Fourth Edition, Oxford University Press (1979) 303-305.