



Windowless helium lamp assisted chemical vapor deposition of hydrogenated amorphous silicon

H. Zarnani, Z. Q. Yu, G. J. Collins, E. Bhattacharya, and J. I. Pankove

Citation: Applied Physics Letters **53**, 1314 (1988); doi: 10.1063/1.100007 View online: http://dx.doi.org/10.1063/1.100007 View Table of Contents: http://scitation.aip.org/content/aip/journal/apl/53/14?ver=pdfcov Published by the AIP Publishing

Articles you may be interested in

Hydrogenation-assisted nanocrystallization of amorphous silicon by radio-frequency plasma-enhanced chemical vapor deposition J. Appl. Phys. **100**, 104320 (2006); 10.1063/1.2390629

Improved hydrogenated amorphous silicon thin films by photon assisted microwave electron cyclotron resonance chemical vapor deposition Appl. Phys. Lett. **74**, 67 (1999); 10.1063/1.122954

Photochemical vapor deposition of hydrogenated amorphous silicon films from disilane and trisilane using a low pressure mercury lamp Appl. Phys. Lett. **48**, 1380 (1986); 10.1063/1.96915

Chemical vapor deposition of hydrogenated amorphous silicon J. Appl. Phys. **59**, 249 (1986); 10.1063/1.336872

Hydrogenated amorphous silicon produced by laser induced chemical vapor deposition of silane Appl. Phys. Lett. **43**, 273 (1983); 10.1063/1.94324



This article is copyrighted as indicated in the article. Reuse of AIP content is subject to the terms at: http://scitation.aip.org/termsconditions. Downloaded to IP: 128.248.55.98 On: Mon, 01 Dec 2014 17:09:55

Windowless helium lamp assisted chemical vapor deposition of hydrogenated amorphous silicon

H. Zarnani, Z. Q. Yu, and G. J. Collins

Department of Electrical Engineering and Engineering Research Center, Colorado State University, Fort Collins, Colorado 80523

E. Bhattacharya and J. I. Pankove

Department of Electrical and Computer Engineering, University of Colorado, Boulder, Colorado 80309-0425, and Solar Energy Research Institute, 1617 Cole Boulevard, Golden, Colorado 80401

(Received 11 May 1988; accepted for publication 28 July 1988)

A windowless helium lamp is employed to assist chemical vapor deposition of hydrogenated amorphous silicon from disilane feedstock gas at a film growth rate greater than 200 Å/min. Material properties in this preliminary study are comparable to the best hydrogenated amorphous silicon (*a*-Si:H) films produced by conventional bulk plasma CVD techniques. The amount of photoconductivity degradation under long-time illumination is more than plasma-deposited *a*-Si-H thin films. Photoconductivity as high as $\sigma_p = 4 \times 10^{-4} (\Omega \text{ cm})^{-1}$ has been obtained.

Hydrogen-free amorphous silicon (*a*-Si) films are not suitable for many optoelectronic devices because dangling bonds, which exist in unhydrogenated amorphous silicon, act as efficient trapping and recombination centers.^{1,2} Hydrogenated amorphous silicon (*a*-Si:H) containing 3–20% hydrogen, however, is highly photoconductive³ and it can be doped *n* or *p* type.⁴ Hydrogenated amorphous silicon has uses in solid-state optoelectronic devices including solar cells,³ image transfer surfaces,^{5,6} and intermediate structures in other solid-state devices.⁷

The most common method to fabricate hydrogenated amorphous silicon (a-Si:H) films is the glow-discharge decomposition of silane⁸ gas. The substrate is immersed in the bulk plasma. The glow-discharge electric field (dc,ac,rf) produces hot electrons of 2–10 eV which subsequently collide with and dissociate input feedstock gases. The electronmolecule dissociation products condense on a heated substrate (200–400 °C) and form an amorphous solid. Note that immersion of the substrate in the bulk plasma brings with it ion bombardment of the growing field which can cause film damage, especially to amorphous superlattice structures.⁹

Film deposition using a helium lamp located remote to the substrate offers different molecular dissociation pathways as compared to direct electron impact occurring in the bulk plasma. For example, in the present research, both vacuum ultraviolet photons and excited helium atoms created inside the lamp participate in collisions with disilane molecules to dissociate it. The role of electron impact dissociation is negligible. Although these same photon-molecule and atom-molecule dissociation processes occur in a bulk plasma, they do not dominate over electron impact dissociation.

Photodecomposition of disilane, ¹⁰⁻¹² trisilane, ¹⁰⁻¹² and also monosilane¹³ has been obtained using monoenergetic excimer laser radiation but the photon absorption cross sections are small in the spectral region where excimer lasers operate (157–200 nm).¹³ Hence, lasers are employed because of their high photon flux which allows for multiple photon absorption.¹⁴ Broadband vacuum ultraviolet lamp assisted chemical vapor deposition^{15,16} and monoenergetic mercury sensitized assisted CVD¹⁷ of disilane to deposit *a*-Si:H thin films have also been reported. In the latter case, excited mercury atoms (Hg^{*}) are used to dissociate disilane in atom-molecule collisions. Disilane has a significant photon absorption cross section in the vacuum ultraviolet (VUV) between 110 and 170 nm.¹³ Therefore, a VUV lamp with significant output between 110 and 170 nm could be used for the direct photodeposition of *a*-Si:H thin films.

Conventional VUV lamps use enclosing walls both to confine the discharge and to maintain the required operating pressure. A window is used to extract the plasma generated VUV photons. VUV window materials are very expensive



FIG. 1. Dependence of 121.5 nm He II line intensity measured near the substrate: (a) vs lamp discharge current and (b) vs lamp helium pressure.

3 0003-6951/88/401314-03\$01.00

© 1988 American Institute of Physics

This article is copyrighted as indicated in the article. Redse of AIP content is subject to the terms at. http://scitation.aip.org/termsconditions. Downloaded to IP. 128 248 55 98 On: Mon. 01 Dec. 2014 17:09:55 and they degrade with prolonged exposure to either moisture or VUV light. VUV entrance windows must also be used on the photo-CVD reactor and undesired film deposits will occur on the inside of the reactor window. Changes in window transmittance vary the intensity of transmitted VUV light from the lamp into the reactor volume and is an undesirable feature.

We utilize an in situ VUV helium lamp¹⁸ that is windowless and provides strong VUV light from the He II spectrum as well as a diffusion flux of excited helium atoms (He*). The intensity of He I VUV radiation from the confined helium plasma reaching the remote substrate is low because it is strongly absorbed by the atomic helium ambient via resonant processes. The He II VUV spectrum is propagated far from the confined helium plasma since the ground-state ion and ion metastable density is low outside the confined plasma and absorption is negligible. Both He II VUV photons and He metastable species are used to dissociate Si₂H₆ molecules in the present apparatus via photon-molecule and atom-molecule dissociation processes. The lamp is placed 14 cm from the substrate, so that any plasma electrons are thermalized to room temperature before reaching the substrate. Hence electron impact dissociation plays no role. In summary, a ring-shaped cold cathode generates a well-confined disk-shaped helium discharge that emits both He I and He II VUV line radiation. Due to strong self-absorption by ground-state helium atoms but not ground-state ions, the VUV resonance radiation emitted from the disk plasma spectrum is selectively attenuated as it travels from the lamp to the substrate. This reduces the He I VUV spectrum impinging on the substrate region but not the He II VUV spectrum. For example, strong He II line radiation at 121.5 nm $(4d^2D - 2p^2P)$ was measured in the substrate region by a UV monochrometer (ARC model VM-502), while the He I resonance line at 60 nm was weak. The dependence of the 121.5 nm He II line intensity versus lamp current is shown in Fig. 1(a). Figure 1(b) shows the VUV intensity versus helium pressure. We find, as described below, that the optimum deposition conditions for a-Si:H thin-film deposition occur when the 121.5 nm He II line emission is strong suggesting a photodissociation role for 121.5 nm radiation.



FIG. 2. Schematic diagram of the two-chamber helium lamp-CVD apparatus.

1315 Appl. Phys. Lett., Vol. 53, No. 14, 3 October 1988

TABLE I. Typical helium lamp-CVD deposition parameters used to form undoped a-Si:H thin films.

Energy source	121.5 nm He 11 line and atomic			
(He discharge lamp)	helium metastables			
Current in ring discharge	0.5 A			
Voltage on ring discharge	600 V			
Donor gas	Disilane (Si_2H_6)			
Flow rates of gases used	Si_2H_6 , 20 sccm			
	He, 200 sccm			
Total pressure	1-1.5 Torr			
Substrates	Corning 7059, fused quartz and			
	(100) silicon wafers			
Substrate temperature	$50 ^{\circ}\text{C} < T_r < 400 ^{\circ}\text{C}$			
-				

The two-chamber deposition apparatus consists of a lamp plasma chamber connected upstream of a remote deposition chamber as illustrated in Fig. 2. Si₂H₆ feedstock gas is introduced into the downstream reactor by a ring-shaped shower nozzle located 1 cm above the substrate as shown in Fig. 2. Helium gas flows through the ring-shaped discharge lamp where it is excited and ionized. Helium afterglow products created in the active discharge flow towards the substrate but only long-lived species such as $He(2^{-1}S)$ and $He(2^{3}S)$ metastables are judged to arrive at the substrate. The distance between substrate and the VUV lamp is almost 14 cm so excited He* levels, other than metastables, will relax via photon emission before reaching the substrate. He II VUV radiation from the plasma disk lamp also impinges on the substrate as described above. As a consequence the primary dissociation mechanisms of disilane in the apparatus of Fig. 2 are judged to be VUV photodissociation of disilane and disilane-helium metastable $(2^{-1}S \text{ and } 2^{-3}S)$ dissociation reactions. The relative importance of these two dissociation mechanisms is not known at this time nor their changing role versus reactor or lamp conditions.

We used both crystalline silicon wafers and Corning 705 glass as substrate materials for (a-Si:H) thin-film depositions using the two-chamber apparatus of Fig. 2. Typical deposition parameters used are listed in Table I. The optical, chemical, and electrical properties of helium lamp assisted a-Si:H films have been characterized. The optical parameters of films deposited on transparent substrates were evaluated by conventional transmission spectrophotometry in the $0.3 < \lambda(\mu) < 3.5$ range. The refractive index $n(\lambda)$ and extinction coefficient $k(\lambda)$ of the deposited films have been calculated using the measured $T \operatorname{vs} \lambda$ transmittance spectra using established methods.¹⁰ The calculated refractive index in the nondispersive (dn/dE = 0) region was n = 3.62. The absorption coefficient $\alpha(\lambda) [= 4\pi k(\lambda)/\lambda]$ versus wavelength was also determined. From the $\alpha(\lambda)$ curve we calculated the optical band gap (E_g^{op}) using the Tauc¹⁹ empirical

TABLE II. Material parameters of a-Si:H films deposited at 350 °C.

Optical band gap	$E_g^{\rm op} = 1.76 \mathrm{eV}$
Refractive index	n = 3.62
Hydrogen concentration	$C_{H} = 4.3\%$
Photoconductivity	$\sigma_{\rm p} = 4 \times 10^{-4} (\Omega {\rm cm})^{-1}$
$(AM1 \ 100 \ mW \ cm^{-2})$	r
Dark conductivity	$\sigma_d = 3.5 \times 10^{-9} (\Omega \text{ cm})^{-1}$
Conductivity activation	$E_a = 0.71 \text{ eV}$
•	

I his article is copyrighted as indicated in the article. Heuse of AIP content is subject to the terms at: http://scitation.aip.org/termsconditions. Downloaded to IP: 128.248.55.98.On: Mon. 01 Dec 2014 17:09:55

TABLE III. Comparison of material properties of undoped a-Si:H films deposited with different techniques.

	Laser CVD			Lamp CVD			He lamp CVD
$\sigma_{\rm nh}(\rm AM1)$	1.2×10 ⁻⁵	2×10 ⁻⁴	2×10 ⁻⁶	10-5	10 -5 - 10 -4	6×10 ⁻⁴	4×10 ⁻⁴
σ_D	2.5×10^{-11}	2×10 ⁻⁹	1×10^{-10}	1010	10 ¹¹ -10 ⁻¹⁰	4×10^{-9}	3.5×10^{-9}
Donor gas	Si ₂ H ₆	Si_2H_6	Si ₂ H ₆	Si ₂ H ₆ Si ₃ H ₈	Si_2H_6	Si_2H_6	Si_2H_6
Energy source	193 nm excimer laser	193 nm excimer laser	193 nm excimer laser	184.9 nm	253.7 nm Hg sensitized	254 nm - weak 185 nm no Hg	121.5 nm He II line & atomic helium metastables
Deposition rate	200–300 Å/min	2 Å/s	100-200 Å/min	0.1 Å/s	2 Å/s	15 Å/min	200 Å/min
Reference	10	11	12	15	17	16	Present Work

relation. The extrapolated value of $(\alpha E)^{1/2}$ vs E yields an E_g value for our *a*-Si:H film of $E_g = 1.76$ eV.

Fourier transforms infrared (FTIR) spectroscopy was used to identify both the total hydrogen content and the form of the silicon-hydrogen molecular structure. Hydrogen-silicon bonding was found to be predominately in the Si-H stretching mode for all the *a*-S:H films deposited over the entire range of substrate temperature investigated. The Si-H hydrogen concentration was found to vary from 4 to 8% as the substrate temperature varied from 350 to 300 °C.

Electrical conductivity of the deposited films was measured in the dark (σ_d) and also under illumination (σ_p) using an AM1 (100 mW cm⁻²) lamp. In this measurement we employed the coplanar electrode configuration using two parallel Ag electrodes for electrical contacts. The σ_p/σ_d ratio was measured to be 1.1×10^5 . The temperature variation of the dc electrical conductivity $\sigma(T)$ versus inverse measurement temperature (Arrhenius plot) determined the activation energy for conductivity to be $E_a = 0.71$ eV for a film deposited at a substrate temperature of $T_s = 350$ °C and E_a = 0.9 eV for a film deposited at a substrate temperature of $T_s = 300$ °C. The activation energy decrease is primarily due to the decreasing content of hydrogen in the deposited film resulting in a lower value of energy gap. Detailed material parameters of deposited films at substrate temperature of T_s = 350 °C are summarized in Table II.

Deposited a-Si:H films under illumination for extended periods of time display a decrease of both photoconductivity and dark conductivity as reported by Staebler and Wronski.20 Changes of spin density21 and infrared absorption²² have also been observed. We also examined the influence of long-time illumination on the steady-state photoconductivity for a deposited film at $T_s = 350$ °C using the helium lamp. Photoconductivity before light soaking was σ_p = 4×10^{-4} (Ω cm)⁻¹ while after a 24 h AM1 light exposure the measured photoconductivity dropped to σ_p $= 2.9 \times 10^{-6} (\Omega \text{ cm})^{-1}$. This final value is ten times less than the best results for plasma-deposited a-Si-H thin films. Subsequently we annealed the sample in a N₂ environment at T = 170 °C for 1 h and the measured photoconductivity was $\sigma_p = 9 \times 10^{-4} \ (\Omega \text{ cm})^{-1}$. A comparison between *a*-Si:H film properties deposited with the helium lamp method and with other photo-CVD methods is summarized in Table III.

In all cases Si_2H_6 feedstock gas was used. We are presently depositing *a*-Si:H thin films using both H₂ and He lamps with silane and disilane feedstock gases to compare results to those described herein.

In summary, we have demonstrated a new method to deposit a-Si:H thin films using VUV photon and atomic metastable dissociation of disilane gas. Helium lamp assisted CVD of hydrogenated amorphous silicon provides material properties comparable to conventional plasma CVD methods. Stability of the deposited films does not deteriorate drastically after long-time photon illumination.

This research was supported in part by the Defense Advanced Research Projects Agency, Applied Electron and the Naval Research Laboratory.

- ¹J. I. Pankove, ed., *Hydrogenated Amorphous Silicon, Part A, Preparation and Structure* (Academic, New York, 1984), Vol. 21A.
- ²U. Voget-Grote, W. Kummerle, R. Fischer, and J. Stuke, Philos. Mag. B 41, 127 (1980).
- ³D. E. Carlson, IEEE Trans. Electron Devices ED-24, 445 (1977).
- ⁴W. E. Spear and P. G. Lecomber, Solid State Commun. 17, 1193 (1975).
- ⁵S. Ishioka, Y. Imamura, Y. Takasaki, and S. Nobutoki, Jpn. J. Appl. Phys. Suppl. 22-1, 461 (1983).
- ⁶I. Shimizu, S. Shirai, and E. Inoue, J. Appl. Phys. 52, 2776 (1981).
- ⁷J. I. Pankove, ed., *Hydrogenated Amorphous Silicon, Part D, Device Applications* (Academic, New York, 1984), Vol. 21D.
- ⁸M. Hirose, in *Hydrogenated Amorphous Silicon*, edited by J. I. Pankove (Academic, New York, 1984), Vol. 21A, Chap. 2.
- ⁹B. Abeles and T. Tiedje, in Hydrogenated Amorphous Silicon, edited by J.
- I. Pankove (Academic, New York, 1984), Vol. 21D, Chap. 12, p. 407. ¹⁰H. Zarnani, H. Demiryont, and G. J. Collins, J. Appl. Phys. **60**, 2523 (1986).
- ¹¹Akira Yamada, Makoto Konagai, and Kiyoshi Takahashi, Jpn. J. Appl. Phys. 24, 1586 (1985).
- ¹²A. Yoshikawa and S. Yamaga, Jpn. J. Appl. Phys. 24, 1585 (1985).
- ¹³Uichi Itoh, Yasutake Toyoshima, and Hideo Onuki, J. Chem. Phys. 85, 4887 (1986).
- ¹⁴Toshihiro Taguchi, Masato Morikawa, Yasuyuki Hiratsuka, and Koichi Toyoda, Appl. Phys. Lett. **49**, 971 (1986).
- ¹⁵Ken Kumata, Uichi Itoh, Yasutake Toyoshima, and Akihisa Matsuda, Appl. Phys. Lett. 48, 1380 (1986).
- ¹⁶Y. Mishima, M. Hirose, Y. Osaka, K. Nagamine, Y. Ashida, N. Kitagawa, and K. Isogaya, Jpn. J. Appl. Phys. 22, L46 (1983).
- ¹⁷M. Konagai, MRS Proc. 70, 257 (1986).
- ¹⁸VUV lamp source provided by Applied Electron Corp., Albuquerque, NM; U. S. Patent No. 4 509 451 (1985) and others pending.
- ¹⁹J. Tauc, R. Grigorovici, and A. Vancu, Phys. Status Solidi 15, 627 (1966).
- ²⁰D. L. Staebler and C. R. Wronski, Appl. Phys. Lett. 31, 292 (1977).
- ²¹H. Dersch, J. Stuke, and J. Beichler, Appl. Phys. Lett. 38, 456 (1980).
- ²²N. M. Amer and A. Skumanich, Physica 117&118, 897 (1983).

I his article is copyrighted as indicated in the article. Heuse of AIP content is subject to the terms at: http://scitation.aip.org/termsconditions. Downloaded to IP: 128 248 55 98 On: Mon. 01 Dec 2014 17:09:55