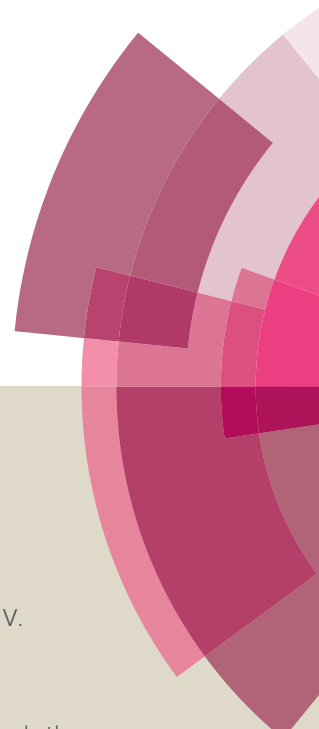
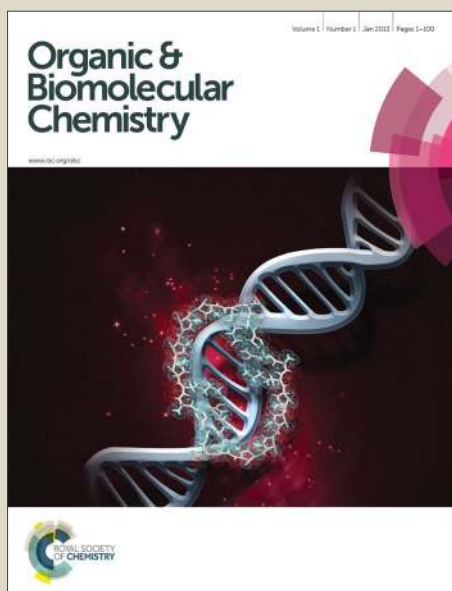


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ARTICLE TYPE

Organocatalytic Asymmetric Decarboxylative Cyanomethylation of Isatins using a Cinchona Derivative Functional Thiourea

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First asymmetric decarboxylative cyanomethylation of isatins is reported here with a thiourea derived from cinchona in good yields and enantioselectivity. This methodology enables the construction of various ethylene substituted hydroxy oxindoles in an enantioselective manner. Enantioselective synthesis of natural products has been accomplished in fewer steps.

10 Introduction

Pharmacologically active scaffolds present in natural products inspire the synthetic chemists to synthesize analogs of the same to increase molecular diversity. Hydroxy oxindole is one such scaffold which received much attention due to its wide biological activity.

15 activity.

Intermediate is a versatile building block to various natural products. The reported methods to enantioselectively synthesize hydroxy oxindoles can be broadly classified into metal catalyzed organocatalytic decarboxylative addition of malonic acid half thioester ketone which is an effective way of forming C-C bond enantioselectively to synthesize intermediate. A recent communication discloses the use of Ytterbium complex of E construction of hydroxy oxindole by Hayashi and coworkers.

25 accomplished the synthesis of intermediate using asymmetric aldol reaction. The resulting intermediates of the protocols require multiple synthetic transformations to accomplish the synthesis of natural products such as 1, etc.

Existence of few methods for synthesizing intermediate stimulated us to develop a methodology for the synthesis of hydroxy oxindoles. Moreover, devoid of nitrogen intermediate demand multiple synthetic steps to total synthesis of natural products. Reduction of a group as shown in intermediate definitely shortens the synthesis of natural products. Allie, et al.

Although decarboxylative addition using malonic ketoacids is well explored, there are only few reports in literature for the enantioselective decarboxylative addition of

cynoacetic acid by achiral version of cyanomethylation of isatins had been reported. To the best of our knowledge, there is no report on enantioselective synthesis of isatins. Herein we disclose the first enantioselective decarboxylative cyanomethylation of isatins with good yield and enantioselectivity using a cinchona derived thiourea catalyst.

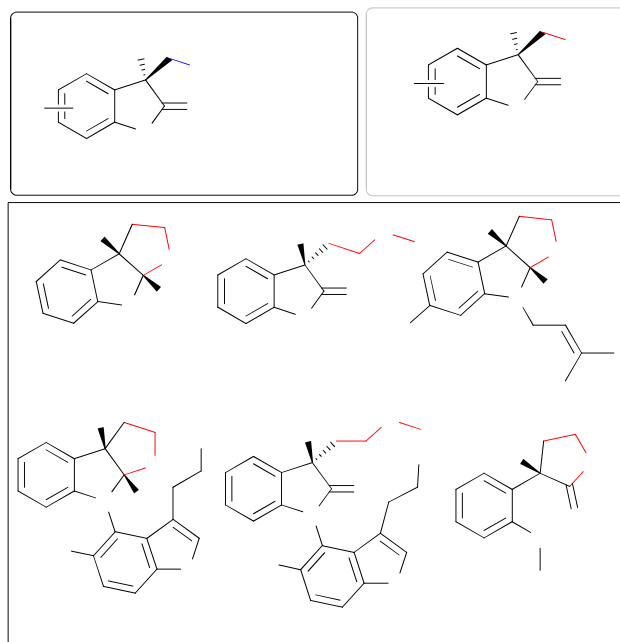


Fig. Selected examples of bioactive molecules which can be synthesized from intermediate in fewer steps

Results and discussion

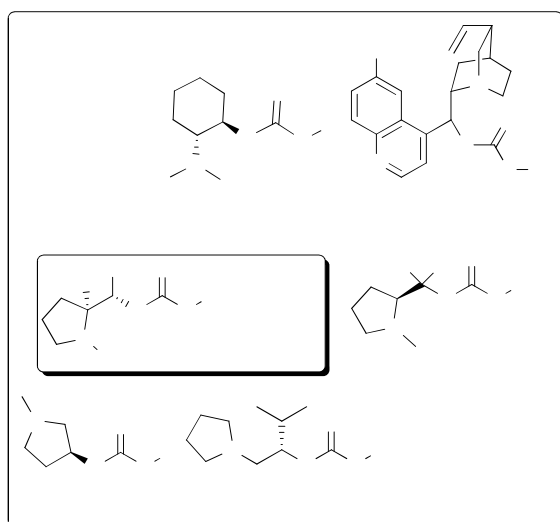
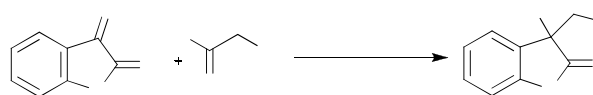
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Electronic Supplementary Information (ESI) available: Table 1, entries 1-5. ICD, these chiral bases catalysed the formation of intermediate.

Although satisfactory yields were obtained, the enantioselectivities were found to be low. In order to identify a suitable chiral base for the reaction, various chiral bases were explored. However, the chiral bases failed to give the expected outcome, we turned our attention to bifunctional organocatalysts containing tertiary amine and thiourea groups. Widely used bifunctional thiourea organocatalysts (4i and 4j) were employed (5 mol %). We hope that a bifunctional thiourea organocatalyst will emerge as an efficient catalyst for the enantioselective cyanomethylation of isatins. In addition, the modular design of thiourea organocatalysts will allow the possibility of altering the functionality at stereogenic centre bearing the thiourea group.

Table 1 Screening of organocatalysts for the enantioselective decarboxylative cyanomethylation of isatins



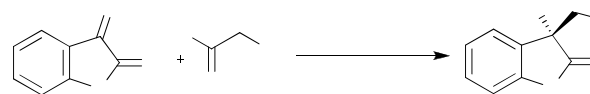
Entry	catalyst	Time (day)	Yield ^a (%)	ee ^c (%)
1	4a	3	78	11(S)
2	4b	3	79	rac
3	4c	3	76	16(R)
4	4d	3	81	07(R)
5	4e	3	79	10(S)
6	4f	3	trace	-
7	4g	3	78	rac
8	4h	3	79	48(S)
9	4i	3	79	57(R)
10	4j	3	79	42(S)
11	5a	3	77	74(S)
12	5b	3	79	80(S)
13	5c	3	79	75(S)
14	6a	5	trace	-
15	6b	3	81	55(R)
16	7	3	trace	-
17	8	3	76	54(R)

^a The reactions were carried out with 1 (0.11 mmol), and catalyst (0.005 mmol) in 0.5 ml of solvent at room temperature. Determined by chiral HPLC.

These results indicate that, identification of a suitable catalyst for

the enantioselective transformation is ongoing. We were delighted to observe that the proline derived bifunctional thiourea organocatalyst 5b efficiently catalyzed the formation of product 3a with good yield and enantioselectivity (Table 1, entry 12). To certain the requirement of newly created stereogenic centre in 5b, a series of thiourea derivatives were employed. More sterically hindered thiourea which has no stereogenic centre at the carbon bearing moiety was unable to catalyze cyanomethylation of isatin (1, entry 14). Although prolinol derived thiourea product 6a in 81% yield with moderate enantioselectivity (55%) was observed (Table 1, entry 15). To prove the value of newly employed thiourea, we evaluated two more thiourea catalysts 7 and 8 that contain alkyl pyrrolidine. Catalyst 7 failed to promote cyanomethylation, product 3a was isolated in 76% yield with low enantioselectivity in the case of thiourea (Table 1, entries 16 and 17). The absolute configuration of product 3a was determined.

Table 2 Optimization studies of enantioselective decarboxylative cyanomethylation of isatins using 5b organocatalyst



Entry	Solvent	mol %	Time(days)	Yield ^a (%)	ee ^c (%)
1	THF	5	5	78	81
2	CH ₂ Cl ₂	5	5	70	33
3	DCE	5	4	trace	-
4	CHCl ₃	5	4	trace	-
5	CH ₃ CN	5	4	trace	-
6	MTBE	5	4	82	86
7	Et ₂ O	5	4	81	73
8	Acetone	5	4	60	44
9	EtOAc	5	4	63	60
10	PhCH ₃	5	4	trace	-
11	DMF	5	6	85	07
12 ^d	MTBE	5	6	trace	-
13	MTBE	2	6	76	75
14	MTBE	10	4	80	80

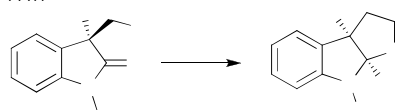
^a The reactions were carried out with 1 (0.11 mmol), and 5b (0.005 mmol) in 0.5 ml of solvent at room temperature. Yield determined by chiral HPLC. Reaction was carried out at 5 °C by organocatalyst 5b and 6b was found to be (S) and (R) respectively (Table 1, entries 15 and 16). The results indicate a suitable catalyst for

that, the newly created stereogenic center at carbon (Table entries 1-7). Various substituted isatins thiourea moiety not only dictates the stereochemical course of the reaction but it is also essential for effective asymmetric induction.

Further optimizations of reaction medium, catalysts in good yields and enantioselectivities for the temperature resulted in only marginal improvements for substrates bearing halogens, electron withdrawing groups and the electron releasing group.

5% of catalyst at ambient temperature and MTBE as the reaction medium were identified as optimized reaction conditions.

the product in 82% yield and 84% ee. Identified suitable functional thiourea which outperformed widely used chiral bases and thiourea catalysts in cyanomethylation of isatin.



Scheme 1 Enantioselective synthesis of CPC

Table 3 Substrate scope of enantioselective cyanomethylation of isatins using organocatalyst

Entry	R ₁ , R	Product	Time (days)	Yield (%)	ee (%)
1	H, H	3a	4	82	84
2	5-Br, H	3b	3.5	78	82
3	5-Cl, H	3c	4	82	84
4	5-F, H	3d	3.5	76	87 (98)
5	5-NO ₂ , H	3e	3.5	78	84 (99)
6	5-OCH ₃ , H	3f	6	72	75
7	5,7-di Br, H	3g	3	78	86
8	5,7-di Cl, H	3h	4	72	90
9	5,7-di Me, H	3i	6	72	81
10	H, Bn	3j	4	75	70
11	H, Me	3k	4	81	80
12	H, <i>tert</i> -butylacetate	3l	4	76	80
13	H, Propargyl	3m	4	80	72

^aThe reactions were carried out in (0.11 mmol), 5 and (0.005 mmol) in 0.5 ml of MTBE at room temperature determined by chiral HPLC. Enantiomeric excess in the parenthesis was obtained after single recrystallisation.

The scope of the reaction was further established by examining various substituted oxindoles as substrates. Isatins were treated with cyanoacetic acid in the presence of 5% catalyst in MTBE at 25 °C. The corresponding cyanomethylated products were obtained in good yields in all the cases. Notably, halogenated substituents are well tolerated in this cyanomethylation and distilled prior to use. Toluene and THF were used in which the respective products were isolated in very good enantioselectivities (Table 3). Reaction proceeded efficiently with good yield and enantioselectivity in the presence of electron withdrawing substituents (Table 3, entry 5). A slight diminish in enantioselectivity was noticed in the presence of electron releasing substituents (Table 3, entry 6). 5-Disubstituted isatins were successfully used as substrates and comparable yields and enantioselectivities of mono-substituted

Reagents and conditions: (i) H₂, LiAlH₄, HCN, CNBH₂, MeOH, r.t.; (ii) H₂, LiAlH₄, HCN, CNBH₂, MeOH, r.t.; (iii) NaH, MeI, dry DMF, yield (over 3 steps).

There are only two reports on enantioselective synthesis of CPC-1. Kitajima employed a Pr₃ mediated asymmetric allylation route to accomplish the total synthesis of CPC-1. Hayashi applied covalent organocatalysis to achieve enantioselective enamine addition to isatin. We demonstrated asymmetric cyanomethylation of isatin as a novel method to synthesize CPC-1 using a covalent thiourea organocatalyst with comparable enantioselectivity in the presence of NaH and MeI and with better overall yield.

Conclusions

In summary, we successfully applied an organocatalytic enantioselective cyanomethylation of isatins for the first time in the literature. It is noteworthy that, the well known systems such as chiral bases, privileged thioureas for this transformation when compared to the previously developed in our laboratory. We further substantiated the importance of substituted oxindole scaffold in the field of medicinal chemistry, we are confident that intermediate II will play vital role in the synthesis of bioactive small molecules. Further exploration of organocatalysis in new enantioselective reactions is being carried out in our laboratory.

Experimental

General remarks

All reactions were carried out in an oven dried flask. Solvents were distilled over CaH₂ and stored under nitrogen. Sodium benzophenone, CH₂Cl₂ and CHCl₃ were distilled over CaH₂. Solvents were distilled over CaH₂ and stored under nitrogen. HPLC was performed on a Shimadzu LC-10AD VP instrument using a 500 MHz instrument using DMSO-*d*₆ as solvent and TMS as reference. Multiplicity follows: s (singlet), d (doublet), t (triplet), q (quartet), m (multiplet), dd (doublet of doublet),

